polymerization of cyclosiloxanes in anion polymerization, started by butyllithium in the presence of DMF

A. Hannioui¹, A. El Malki¹, E M. Rakib¹, N. Knouzi²*, M. Vaultier³

Two methods of polymerization give

Laboratory of Organic and Analytical Chemistry, Faculty of Sciences and Feel polys, iloxanes : the University Sultan Moulay Slimane, ²University Hassan II, Faculty of Science Sidi Othmane, BP6023, Casablanca, Morocco². ³Laboratory of Synthesis and Electrosynthesis Organics UMR 6510, University Rennes I. France³. * Corresponding author. E-mail : hannioui7@yahoo.fr Digomer This type of polymerization was

opening of cycle of the cyclosiloxanes. This type of polymerization was described in many publications ¹⁻⁴. Indeed, the cycles having a grouping Z present a polarized connection likely to open in the presence of a center active.

Our objectivies is to study the polymerization of cyclosiloxanes by opening of cycle. The substituent of the organosiloxanes more used is the methyl group. As we explained previously, we choose to study the behaviour of the cyclosiloxanes carrying groups Alkyls, Vinyls, Alcools... etc, in anion polymerization, started by butyllithium in the presence of an activator DMF, at the temperature of 120°C. After having fixed our choice on the principal objectives, we defined a strategy of the synthesis. It appeared convenient to us to use monomers whose substituents are alkyls, aryls or Alcohols, in order to study the influence of the latter on polymerization. D₄^R ₊ n BuLi

1. T.Tsuruta, Y.Kawakami, understanding, polymerscience vol.3.457, Pergamon Press (1989).

2. D.J. Brunelle, Boxing ring Opening polymerization, 1, Ed.K.J.Ivin, Tsaegusa, Elsevier, London (1984).

3. K.J.Ivin, Tsaegusa, Boxing ring opening Polymerization, 1, Ed.K.J.Ivin, Tsaegusa, Elsevier, London (1984).

4. G.Odian, polymerization principles and application, 582, Polytechnica (1994).