Aminoacid-Asisted Synthesis of TiO2 Nanocrystals with Controllable Shape and Size: A Novel Agent for the Fabrication of MEH-PPV/TiO2 Photovoltaic Materials

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Photovoltaic (PV) devices based on conjugated polymers have become attractive for use in inexpensive large area and low weight devices. However, for practical applications, several factors limit the efficiency, especially the low mobility of charge carriers, particularly electrons. To improve this limitation, nanocrystalline Titanium dioxide (TiO2) with its chemical stability, easy control of size and shape, proper band gap and low cost, is considered as one of the most promising materials for efficient electron transport. The main objective of this work is i) to synthesize TiO2 nanocrystals (NCs) and ii) to optimize the dispersion of these NCs into conjugated Poly[2-methoxy-5-(2'-ethylhexyloxy)-p-phenylene vinylene] (MEH-PPV) polymer in order to develop hybrid MEH-PPV/TiO2 films for photovoltaic cells. Various routes have been developed to synthesize TiO2 nanocrystals through the solvo/hydrothermal reaction of commercial titanium dioxide nanopowders or titanium butoxide in the presence of selective capping agents (e.g., amino acid and decanoic acid) at temperature varying between 100 and 180oC. The various shapes of such TiO2 crystals (including nanorods, nanotubes, microspheres, hollow microspheres, quasi-nanospheres) have been controlled by changing the reaction parameters. In the second step of this work, the effect of the shape and size of TiO2 NCs on the efficiency of MEH-PPV/TiO2 photovoltaic materials will be investigated.